

**الألمنيوم عالي القوة في مقاومة التآكل تحسين فعالية سبائك النحاس الصلب باستخدام**

**لـ**

**حامد صالح مهدي**

**رئيس مهندسين**

**وزارة النفط العراقية – شركة النفط الوطنية**

**شركة نفط البصرة**

**688800**

**هيئة إدارة وتطوير مرافئ التصدير**

## TABLE OF CONTENTS

### APSTRACT

..... 3

Aim of research

..... 3

introduction

..... 4

1.1 Introduction

..... 4

1.2 Key Alloying and Impurity elements

..... 5

1.2.1 Copper

..... 5

1.2.2 Manganese

..... 5

1.2.3 Magnesium

..... 6

1.2.4 Silicon

..... 6

1.2.5 Iron

..... 6

1.2.6 Tin

..... 7

1.2.7 Titanium

..... 7

1.2.8 Nickel

..... 7

1.3 History of Casting Al-Cu alloy

.....7

### 1.3.1 Al-Cu phase diagram

8

#### 1.3.1.1 B206

alloy.....

8

#### 1.3.1.2 Solidification reactions of 206.2 alloy

8

### 1.4 Casting Technology

.....9

#### 1.4.1 Permanent Mold Casting Process

9

#### 1.4.2 Hot-cracking of Al-Cu-Mn-Mg alloy3. discussing figures and result

9

##### 1.4.2.1 Effect of Silicon on the fluidity and hot cracking of Aluminum-Copper alloy

10

##### 1.4.2.2 Fatigue crack growth in cast Al-Cu alloy A206 with different levels of porosity

10

### 1.5 Heat Treatment of Al-Cu alloys

11

#### 1.5.1 Solution Treatment

11

#### 1.5.2 Precipitation hardening

12

#### 1.5.3 Natural Aging

12

#### 1.5.4 Artificial Aging

(AA).....

13

##### 1.5.4.1 Single stage Artificial Aging (AA)

14

##### 1.5.4.2 Multi-stage Artificial Aging (AA)

15

#### 1.5.5 Effect of Si on the microstructure and mechanical properties of Al-4.5%Cu alloys

16

#### 1.5.6 Effect of Cu and Si on mechanical properties in Al-Cu-Si-Mg alloys

17

#### 1.5.7 Chemistry/Property relationship in AA206

alloy..... 18

#### 1.5.8 Mechanical properties of B206 alloy

20

<b>1.6 Corrosion of Al-Cu alloys</b>	<b>21</b>
.....	
<b>1.6.1 Immersion testing of heat treated 201 alloys in NaCl/H<sub>2</sub>O<sub>2</sub> corrosive solution</b>	<b>22</b>
.....	
<b>1.6.2 Stress Corrosion Cracking</b>	<b>23</b>
.....	
<b>1.6.2.1 Intergranular stress corrosion behaviour of Al-Cu-Mg-Ag alloys</b>	<b>24</b>
.....	
<b>1.6.2.2 Pitting corrosion of Al-Cu alloys</b>	<b>24</b>
.....	
<b>.conclusion</b>	<b>25</b>
.....	
<b>.recommndation</b>	<b>26</b>
.....	
<b>. references</b>	<b>26</b>
.....	

## APSREACT

B206 (up to 5 wt percent Cu) is the foundry amalgam of aluminum that is the most grounded in use today. In a number of car applications, B206 combination can be applied to decrease vehicle weight, such as suspension knockers and vehicle control arms. Within the 206 combination family, the disposal of hot tearing stirred the fascinated. In any events, the amalgam B206 is vulnerable to intergranular erosion/pitting which restricts present uses.

Heterogeneous spreading of intermetallic Cu-containing accelerates in the as-fast state occurs in severe intergranular erosion. The 3-step ST+2-step AA is superior to the 2-step ST + 1-step AA erosion resistance. AA time, which was dispensed with intergranular erosion but resulted in corrosion on the moo level. The extension decreased when the AA temperature and time were increased. It is challenging for the overaged state to get both an incredible erosion resistance and an extension (510%).

## Introduction

Al-8% Cu amalgam, known as the 12 combination, was the predominant aluminum casting amalgam used in America. For vehicle cylinders and cylinder heads an Al-10 percent Cu combination (no. 122) was used. Alfred Wilm discovered in 1909 that a combination of Al-4.5% Cu-0.5% Mn might be strengthened if the temperature was extinguished after the maturity. The amalgam was known as 'Duralumin' and formed the concept of today's Al-Cu family of combinations. The Al-Cu casting amalgams offer excellent mechanical qualities, high temperature quality and low cycle weakness, but are difficult to cast, mainly since they are heated. Over a long period, founders found increased castability of other components, especially Si. The Al-Cu amalgam family was mainly replaced by Al-Cu-Si and Al-Si-Mg amalgams thereafter.

Grain finishing minimizes the risk to heat fracture in net-shaped casting. It is well recognized for a long time. A 1970 study by Davies, who found out that the total length of warm tears was proportional to the average grain estimate, could be the meditation which appears most obviously. Ti was incorporated in alloys for Al-Cu casting grain refining in the early 1930's. There was a huge increase in castability and this quickly became the established honing. The most excellent founders found

The TiAl<sub>3</sub> Intermetallic Solvency Constraint of pure aluminum (littlest) grain sizes occurred in concentration over 0.15 wt percent Ti. This was recognized as the chemical composition restrictions for the 201, 202, 204 and 206 amalgams in the Aluminum Affiliation: A minimum 0.15 percent Ti material is shown For the termination of warm tears in A206.0 and A535.0 (Al-), the combination of grain refinement and thermal administering of the form temperature is crucial (6.2-7.5) Mg-(0.1-0.25) Mn-0.15max. Si-0.15max. 0.05max Fe-. Amalgams Cu—(0.1 –0.25)Ti)

The Al-Casting AA 206 family (up to 5% Cu) has mechanical characteristics that are near to those of the bendable press, e.g. up to 25% extensions; In addition these alloys are essentially restricted to military and aviation applications, since they are prone to hot shortening, as candidates for vehicle suspension applications have been explored. Refining of grains has advanced the castability of this alloy by reducing its sensitivity to warm wear.

B206 could be a combination enlisted that benefits from a disclosure relating to Ti grain finishing. Since it's defective to the inter-granular stretch erosion that is harmful to the cast component's long-cycle execution, B206 amalgam car applicationen are now limited. The high temperature counterfeit maturing and lengthy ripening duration can control intergranular corrosion (T7 mood assignment). However, the overwhelmed

condition will result in the combination having less mechanical qualities. Thus, the B206 is particularly concerned with a warm treatment that equalizes both corrosion resistance and mechanical qualities combination.

Warm Treatment of the B206 amalgam should be improved to ensure that the al-Cu phases in the course of arrangement therapy (ST) are homogeneous and the Al-Cu phases are uniformly dispersed in fake maturing therapy. The erosion behavior was mainly investigated utilizing immersion tests and potentiodynamic tests on tests that have undergone several warm-thermal treatment plans, including a 2 or 3-step arrangement and varied maturation plans (normal and counterfeit). The heat treatment techniques which generate the optimal mechanical characteristics do not basically produce the most safe materials for erosion. In case B206 is to be utilized in automotive applications such as suspension components, at that point tall ductility (lengthenings  $>7\%$ ) is required for crashworthiness .

## **Aim of research**

This study aims most at creating a warm condition of treatment that gives both incredible erosion resistance (no intergranular erosion) and mechanical properties (270 MPa Surrender Quality (YS), 310 MPa Extreme Malleable Quality (UTS) and a minimum of 10% add up to an extension of amalgam B206). The hot spots are as you would like them to be The following are the detailed aims:

- 1- Potentiality polarization tests and tribocorrosion tests for warm treated examples are performed to test drenching erosion.
- 2- Characterizing the as-casted and warm-treated AA 206 eroding and mechanical properties combination tests.

Comparison of warm-treated trials with and without common maturation involves -3 two steps vs. three steps arrangement treatments (ST) and one step versus two steps of (( artificial maturation (AA) (NA

## **Key Alloying and Impurity elements .2**

### **Copper 2.2.1**

In B206 casting amalgams copper is the most alloying component. Because of its moderately high aluminum dissolvability and its refinement effect, it is one of the most important alloy components for aluminum. For copper additions of 4 to 6%, depending

on the impact of other components, combination reinforcement would be the best. Copper provides significant precipitation hardening features and fabulous lifted temperature characteristics for amalgams as a central alloying components in aluminum. Copper is regularly used to speed up and increase the ages by magnesium, which solidifies at room temperature

### **Manganese .2.2.2**

Manganese in essential aluminum can be a common contamination. It can be shown up to 1 percent in numerous Al-Cu alloys and can enter the (Al) strong structure after solidification. In order to relieve the detrimental effects of contaminations such as printing press and silicon, manganese is commonly used in Al-Cu In order to reduce the embrittling impact, the manganese is also included. It is combined with press and silicone and increases solid quality. Manganese expansion leads to a more general magnesium precipitation and allows a magnesium moo substance. The proximity of .Fe, Si and Ni could lead to the arrangement of eutectic root mn-bearing stages The 0.25 per cent expansion of Manganese does not clearly have a negative effect on the maturing of binary amalgams of Aluminum-Copper, but the level of aging decreases by 0.5 to 1.0 per cent. The undissolved component contains copper and manganese in 0.5%-1.0% manganese combinations. Its proximity would reduce the amount of copper in a strong position. This is often why precipitation with these ..amalgams decreases at room temperature

### **Magnesium 2.2.3**

The reaction of aluminum-copper amalgams to heat treatment includes magnesium. Increased quality after treatment and extinguishing of aluminum-copper amals with magnesium. The subgrain size is expanded and the separation thickness decreases with expanding magnesium substance. Magnesium forms the  $Al_2CuMg$  stage that accelerates at  $505-507^{\circ}C$  as part of the ternary eutectic (Al)+ $Al_2Cu$  + $Al_2CuMg$  and increases the solidification range considerably] in the course of non-equilibrium cementing

### **Silicon 2.2.4**

After pressing, silicone is the most significant pollution levels (0.01 to 0.15 percent) in electrolytic commercial aluminium. Silicone is included in Al-Cu combinations to make it easier to cast and reduce the heat. As a result, Al-Cu amalgams have always been replaced with Al-Cu-Si alloys. However, silicone expansion significantly

decreases the combination ductility. It is not copper-forming and usually has high solvency (Al). This improves casting, but Si is horrible for warm strength in amalgam. In a double aluminum-copper amalgam the nearness of 0.25 percent silicone had no effect on aging properties.

### 2.2.5 Iron

Manganese, nickel and copper are the iron stages. In Al-Cu-Si combinations it is the central depletion. An increase of the press material affects the mechanical properties, in particular the durability of the breaking (stretching) and other characteristics. The sensitivity to the proximity of this component is less castable. The age-hardening sum of room temperature decreased in proportion to the amount of iron present in an Al-Cu parallel combination and the maturing period was completely eliminated by an increase of up to 1.5percent in press.

### 2.2.6 Tin

Trace element or miniature Sn alloy increments are well known to prolong Al-hardening Cu's response at a temperature of lifting (130 °C and 190 °C) maturity. In an amazingly finest and uniform dispersion of the 6' precipitation, Ternary Sn addition comes to reinforce the Al-1.7% Cu combination.

### 2.2.7 Titanium

Titanium is considered a grain refiner, and the reduction in the grain estimate is exceptionally compelling. In addition, it creates far more; much better; stronger; better">a far better dispersion of insoluble ingredients, porosity and non-metallic inclusions which results in a critical change of the material's mechanical properties. As the grain estimate controls the porosity dispersion of the components, the mechanical properties of the amalgams are extremely sensitive to the size of the grain. With AlTi5B1 expansion, the cementing structure is adjusted and the grain is estimated to be smaller.

### 2.2.8 Nickel

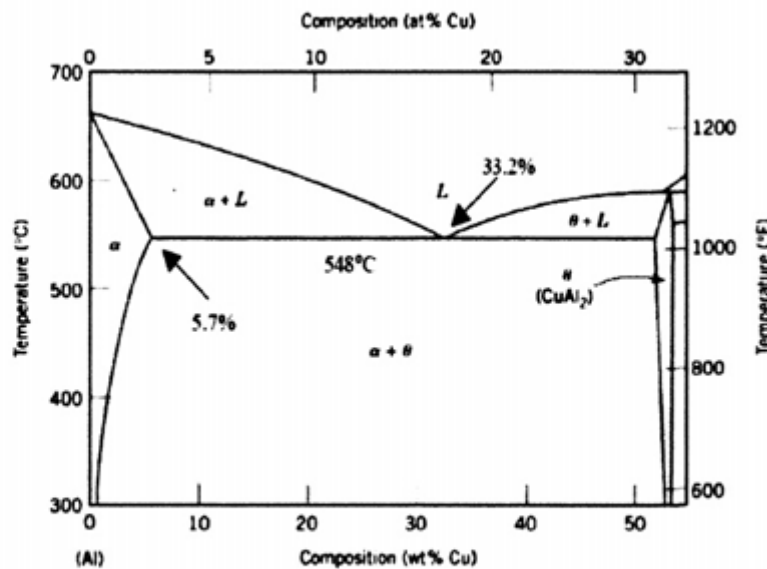


Al<sub>6</sub>Cu<sub>3</sub>Ni or Al<sub>3</sub>(Cu, Ni)<sub>2</sub> nickel of copper shapes. The compounds remain thermally stable and can advance mechanical characteristics at high temperatures. They reduce the concentration of copper in aluminum, however.

## 2.3 History of casting Al-Cu alloys

### 2.3.1 Al-Cu phase diagram

The powerful aluminum terminal in the twin aluminum-copper frame balances with the intermetallic stage 6 Al<sub>2</sub>Cu, because there is some strong solvency. The eutectic phases which are isolated from the fluid in the hardened response are CuAl<sub>2</sub>, an intermetallic stage of 53.5% Cu and the strong aluminum arrangement of 5.65% Cu (1)



**Figure 2.1: Aluminum rich portion of the Al-Cu phase diagram ( 1)**

#### 2.3.1.1 B206 Alloy

The copper content of the B206 alloy is between 4.5% - 5.5%. Table 2.1 provides the composition of B206 alloy: Table 2.1: Alloy composition B206.0 registered

Table 2.1: Registered B206.0 alloy composition

Cu	Mg	Mn	Fe	Ti	Si	Zn	Others	Al
4.2-5.0	0.15-0.35	0.2-0.5	0.10	0.10	0.05	0.10	0.15	balance

### 2.3.1.2 Solidification reactions of B206 alloy

The solidification sequence of the B206 alloy is summarized in Table 2.2 .

Table 2.2; Formation of various phases during the solidification process in a B206 alloy

Reactions	Temperature (°C)
Development of dendrite network	651-649
Liquid $\rightarrow$ (Al) + $Al_6(MnCuFe)$	649
Liquid + $Al_6(MnCuFe) \rightarrow$ (Al) + $Al_{20}Cu_2Mn_3$	616
Liquid $\rightarrow$ (Al) + $Al_{20}Cu_2Mn_3$ + $Al_2Cu$ + $Al_7Cu_2Fe$	537
Liquid $\rightarrow$ (Al) + $Al_2Cu$ + $Al_2CuMg$ + $Mg_2Si$	512

Arrangement of consistent dendrites at about 10-15 ° C below liquid temperature and at about 0.30-0.35 strong division throughout the hardening process. Within an interdendritic locale of over 170-180°C, the remaining division liquid (0.65-0.70) has cement. B206 amalgam casts are therefore extremely helpless to heat tearing

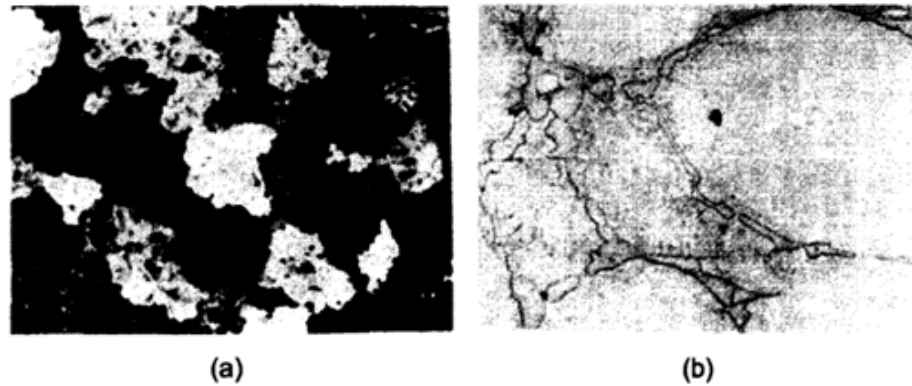


Figure 2.2: Microstructure of B206 alloys (a) grains (b) interconnected dendrites

## 2.4 Casting Technology

### 2.4.1 Permanent Mold Casting Process

Permanent mold refers to the tools used for aluminum castings. The moulds or dies are usually made from high amalgam machines or steel (both exceptionally thick). A slender layer of safe and warm material, such as clay or sodium sulfate, covers the depressed surfaces of the shake. At that point are gathered the metal moulds which comprise two or more parts. The molds are preheated at a given temperature and an extremely near warm adjustment is maintained through water cooling and other radiation procedures.

At this point the fluid metal is best shaped into a spruce. Through the weight and speed initiated by gravity the metal flows into the form cavity. If the metal is cemented, the shape is opened and cast out. In comparison to sand casting processes, the preferences for unchangeable casting handles are far better wrapping, accurate and consistent dimensional control and enhanced mechanical properties. Critical reserve vitality funds can be achieved through permanent casting forms with sand and/or permanent mouths, using gravity and/or low-pressure. Vitality reserve funds of 250kWh are conceivable per ton of castings by switching from sand to changeless shape .

### 2.4.2 Hot cracking of Al-Cu-Mn-Mg alloy

#### 2.4.2.1 Effect of Silicon on the fluidity and hot cracking of Aluminum-Copper

Weijing et al. explained that the Si expansion can advance the Al-5.0Cu-0.4Mn-0.3Mg combination's smoothness and hot cracking propensity. The length of the winding form can be measured with ease. The fluidity property of the amalgam increases from the beginning and decreases from 1 to 2 percent at that point. As illustrated in Figure 2.3 (a), after expanding 2% Si content, his smoothness property increases again. The ease of use was tested through winding, and the hot division was tested with hot crack ring shape. The propensity of heat splitting decreases significantly as Si substance increases:

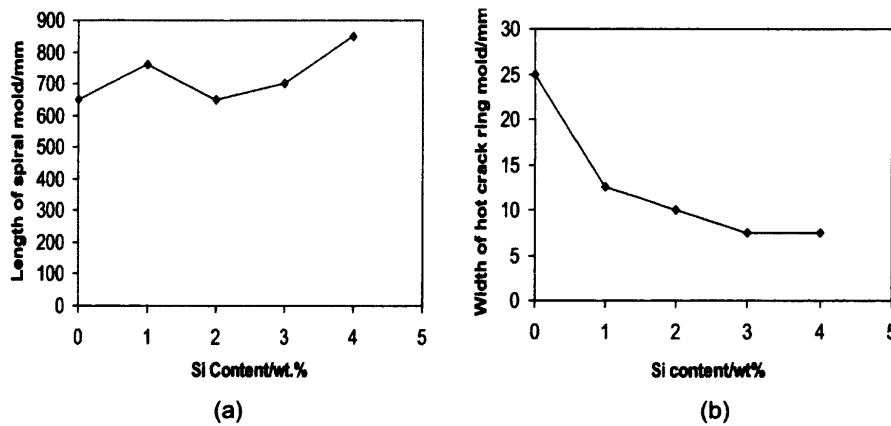


Figure 2.3: a) Variation of fluidity with different contents of Si. b) Variation of hot cracking tendency with different contents of Si

#### 2.4.2.2 Fatigue crack growth in cast Al-Cu alloy A206 with different levels of

##### Porosity

Porosity is an imperative cast surrender that affects the mechanical characteristics of the casting in a coordinated way. Cast amalgams contain regularly large, small pores from either interdendritic (more often not sporadic) or captured (circular) hydrogen and other gasses.. Ambalgams regularly contain small and large pores. The A206 amalgam is the highest virtue in the 206 series. It is usually used in aircraft applications. The A206 was also used in car components such as knuckle suspensions. The porosity impact on weariness partition for two distinctive features was investigated by Rading et al.

The consideration secured administrations both close and mid-range. There has been no accurate relation (chilled or unchilled) between casting conditions and the normal level of porosity. They also found that weariness divided growth rates, especially close to the threshold system, with increased levels of porosity. The process of hot isostatic pressing (HIP) reduces porosity from 4% to below.4%. But its effect on fatigue crack growth is confined to the close threshold system

## **2.5 Heat Treatment of Al-Cu alloys**

Heat treatment is mainly done to ensure that the aluminum casting combinations have the optimal combination of ductility, strength and erosion resistance. Heat treatment includes all warm hones in its broadest sense, with the aim of adjusting the metallurgical structure of the casting to control the changes of physical and mechanical features to meet specific design criteria. The steps included in the Al-Cu combination heat treatment are:

- Solution Treatment (ST),
- Quenching and
- Natural/Artificial Aging.

### **2.5.1 Solution Treatment**

Solution treatment is planned to maximize the solvency of components interested in any consequent ripening therapy. It is most convincing when there is the greatest solvency close to the solid and eutectic temperature. Early softening must be avoided in any case of low temperature eutectics. The temperature over the solvent temperature should be the highest. Treatment Arrangements (ST) do not dissolve intermetallic copper rich components into the Aluminum network at moo temperatures

During Solution Treatment (ST), Cu and Mg are put in a solid solution and the structure becomes uniform. The dissolved copper into the aluminum produces the desired super saturated solid solution when the solution is finished with the copper (rapid cooling).

It was detailed that  $Al_2Cu$  can contribute to the reinforcement of the Al-Cu combinations by strong solutions and by strengthening momentum. The concentration of manganese in casting alloys containing 5% Cu during hardening can reach 2% within the super-saturated strong solution. Arrangement of eutectic nonequilibrium (Al) +  $CuAl_2$  and supersaturated Mn in (al) arrangements causes harmonic deviation during solidification. Manganese is precipitated only when copper reaches the desired concentration within the solid solution

### 2.5.2 Precipitation hardening

Precipitation solidification can occur if the phase is dissolved at a higher temperature, but with a declining temperature it is strongly dissolved. Includes the design of consistent clusters of solvent particles as part of the reinforcement process after arrangement treatment. The arrangement in a precipitation hardening framework like Al-Cu for a coherent acceleration takes place in several steps. The al-Cu alloy includes solvent insulation districts

Or clustering after a solid solution is extinguished. This nearby clustering was initially recognized and named Guinier and Preston as the GP zone. Extensive copper moloculate clumps on {100} matrix planes were formed with extra maturity. This is GP2 or 0" in structure. Next, certain  $CuAl_2$  or 0' platelets on the matrix plane {100}. The balance phase  $CuAl_2$  or 0 is formed with a further aging phase. High proportions of 0 fall through increase in time and temperature to soften the alloy.

### 2.5.3 Natural Aging (NA)

Natural maturation could be an uncompromising handle of precipitation which occurs at room temperature for days and weeks following arrangements (ST). At room

temperature the abundance of super-saturated aluminum is increasing progressively. Depending on the combined frameworks, Solute atoms are either clustered or segregated to selected nuclear grid planes, making GP areas safer for separation development by cross section and thus more rooted.

Abis et al[35] observed early stages of transformation during natural aging of a metastable Al<sub>4.4</sub>Cu<sub>1.7</sub>Mg alloy (NA). At 500 °C and cold rolling up to 1,5 mm thickness, the alloy is homogenised for 10 h. It then is treated at the temperature of the room at 520°C for 1 h and water is extinguished. The formation of Guinier-Preston-Bagariastkij-Al<sub>2</sub>CuMg (GPB) + Guinier-Preston-Al<sub>2</sub>Cu (GP) zones suggest DSC, TEM and corresponding initial hardening increases.

#### **2.5.4 Artificial Aging (AA)**

Artificial maturation could be a preparation to heat casts at intermediate temperatures in the soaked condition. It accelerates the transitional (metastable) forms of balance of a certain amalgam context. These transitional speeds remain consistent with the strong network and contribute to the strengthening of precipitation. Ductility depends heavily on the falsified maturation. In the asquenched condition, ductility is large and decreases with an elevated solidification temperature.

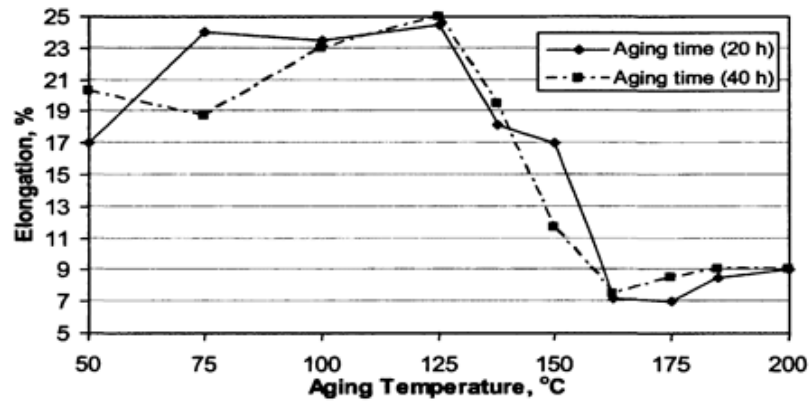
##### **2.5.4.1 Single stage Artificial Aging (AA)**

The reinforcement handle at a particular temperature is a single step AA. It is frequently assumed that the microstructure and mechanical properties remain unchanged at a lower temperature for a non-conclusive period after the aluminum combined is misleadingly developed at mid-road temperature (e.g. T6 at 190°C). In any case, later perceptions of Al-Cu-Mg amalgam with a much lower solvent substance have shown that opening at room temperature may remain multi-faceted after such amalgams have first matured in 180°C, thus encouraging solute dissemination leading to precipitation (assistant) at room temperature.

Meissner[39] has investigated the effect on duralumin of natural+artificial aging (4.2 percent Cu, 0.5 percent Mg, 0.25 percent Mn, 0.3 percent Fe, and 0.3 percent Si). The

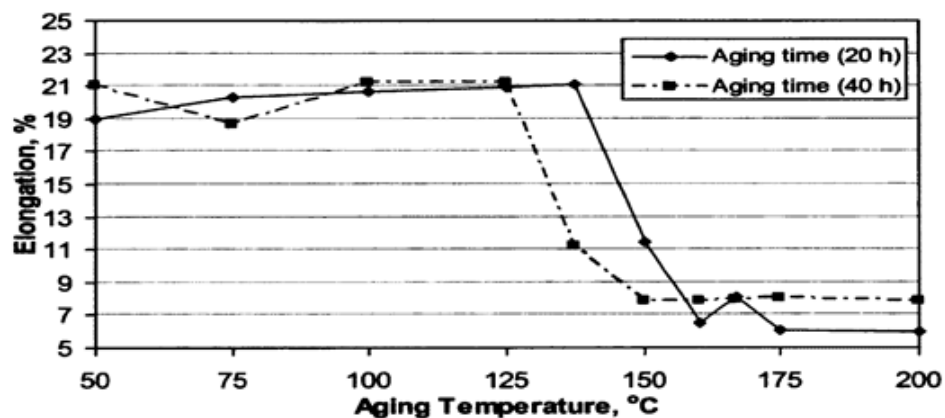


temperature of artificial ageing ranged from 50°C to 200°C. It took 20 hour and 40 hours to reach the Artificial Ageing (AA)



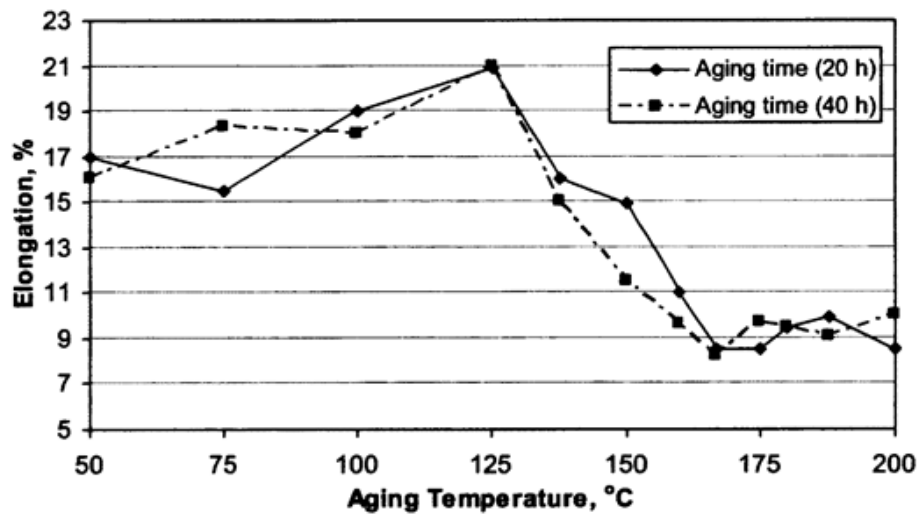
**Figure 2.4: Effect of artificial aging on elongation for duralumin (artificial aging was applied immediately after quenching from about 510°C**

Figures 2.4 and 2.5 show almost the same properties as the extension compared to the aging temperature. This shows clearly that common maturation has no impact on duralumin. The cold-rolling treatment offers a high yield, combined with high malt quality and a large stretch



**Figure 2.5: Effect of artificial aging on elongation for duralumin (artificial aging was applied after complete age-hardening at room temperature for 5 days**





**Figure 2.6: Effect of artificial aging on elongation for duralumin (artificial aging was applied after age-hardening at room temperature and cold-rolled)**

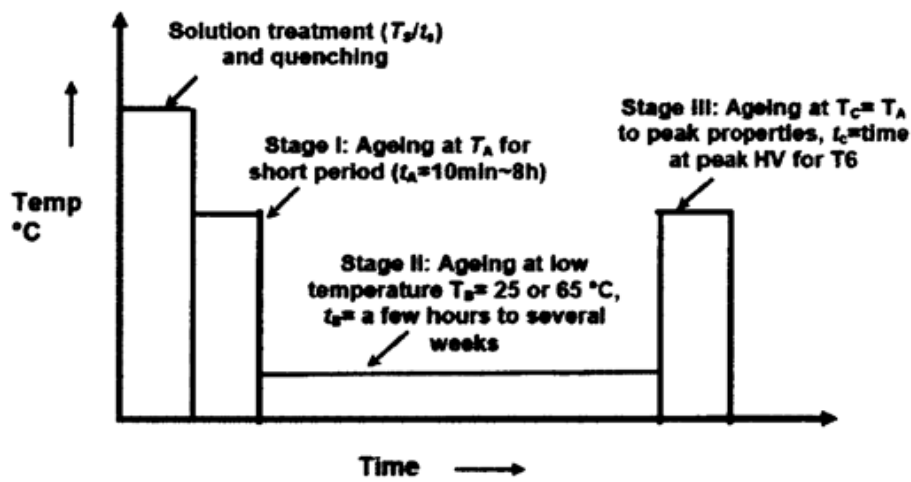
#### 2.5.4.2 Multi-stage Artificial Aging

As the multilevel maturing medicines Lumley, Polmear and Morton are famous for, aluminum combination properties are regularly used. A well-known case is the T73 mode in which Manufactured Maturing takes place at the next temperature (160°C) after a moment's treatment. Another duplex medication is to age naturally after and before artificial aging at room temperature.

Another multi-stage maturity drug framework, particularly obstructed aging therapy, (assigned T6I6), was detailed later, with the assertion that further changes to the quality are exceptionally compelling through standard T6 therapy. That includeth hindering the treatment of T6 by removing the alloy taken for a certain period of time before the treatment of T6 after maturing to lower temperatures (25-65 ° C) until the hardness of the crest is reached.

Goh et al examined the impact on age-hardening responses of aluminum amalgams of multi-stage maturing medicines. They detailed that AA6111, AI-4.0Cu-0.8Mg, and AI-4.0Cu-0.8Mg-0.8Si combinations have little effect although they are refined in accelerated distribution. Gao et al detailed on a 3-stage T6I6 hindered maturing treatment (see Figure 2.7). Interrupted maturing comes about in nucleation of finely

Dispersed speeds within the last aged microstructure. Tensile strength, hardness and durability of the combinations submitted to the T6I6 treatment increased between 5% and 30% depending on its composition and special preparedness conditions



**Figure 2.7 Schematic representation of the three stages of T6I6 interrupted aging**

**treatment. The selection of a long initial Stage I and a suitable Stage II aging time is important. In the most cases, the Stage II aging at 25°C shows a more beneficial effect than aging at 65°C**

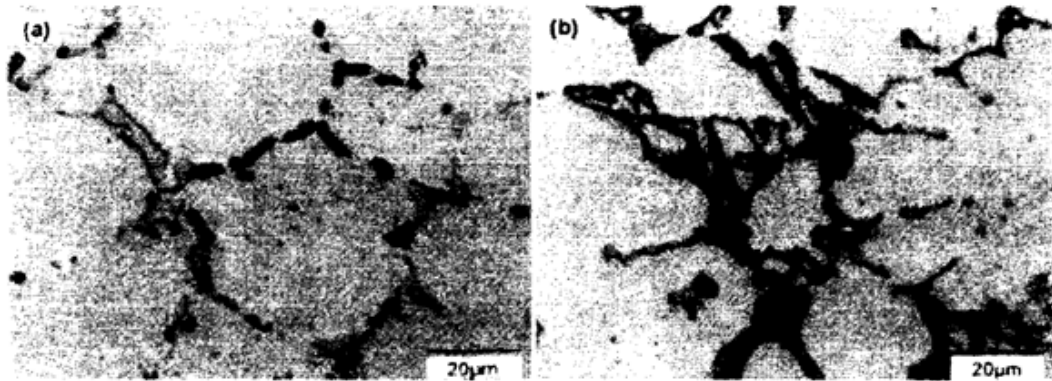
### 2.5.5 Effect of Si on the microstructure and mechanical properties of Al-4.5%Cu

#### Alloys

Han et al detailed the impact on Al-4.5% Cu cast combinations of a Si expansion. Specifically, they used four combinations: Al-4.5% Cu, Al-4.5% Cu-1% Si, Al-4.5% Cu-2% Si and Al-4.5% Cu-3% Si Specifically (wt percent ). At the grain limits of the ash-filled amalgams the coarse-like Si stages have been dispersed: Fig. 2.8. If depends upon its morphology and dispersion, the impacts on the mechanical properties of alloys

The extending decreases with an increase in si substance under the as-cast condition. Al-4.5% Cu-2% Si amalgam has the optimum mechanical properties after T4 warm

treatment. This may be caused by the change in circular particles of the plate-like Si. Fine Si circular particles are speeded up after extinguishing of Al- 4.5%Cu-2%Si amalgam



**Figure 2.8: Microstructure of the as-cast alloys: (a) Al-4.5%Cu-1%Si (b) Al- 4.5%Cu-2%Si**

**Table 2.3: Mechanical properties of as-cast Al-4.5%Cu alloys with Si addition**

<b>Alloy</b>	<b>Tensile Strength, MPa</b>	<b>Elongation, %</b>	<b>Hardness, HB</b>
Al-4.5%Cu	187	10.6	51.8
Al-4.5%Cu-1%Si	181	7.2	54.8
Al-4.5%Cu-2%Si	179	6.4	57.1
Al-4.5%Cu-3%Si	184	6.0	58.4

**Table 2.4: Mechanical properties of the alloys (as in Table 2.3) after T4 heat treatment**

<b>Alloy</b>	<b>Tensile Strength, MPa</b>	<b>Elongation, %</b>	<b>Hardness, HB</b>
Al-4.5%Cu	244	11.8	64.9
Al-4.5%Cu-1%Si	232	7.8	60.8
Al-4.5%Cu-2%Si	313	11.6	78.4
Al-4.5%Cu-3%Si	232	5.6	70.2

### 2.5.6 Effect of Cu and Si on mechanical properties in Al-Cu-Si-Mg alloys

The effect Cu and Si on 1, 3, 4.5, 6 wt. and 0, 5, 7, 12 and 18 wt.% silicon was examined by Zeren[44]. The range of Mg varied between 0.90 and 1.06% wt.

The solutions were treated 4 hours at 490 degreesC, then aged at 180 degreesC, 5,10,15, 20 hours. The solution was treated in water.

**Table 2.5: Effect of Cu content of Al-Cu-Si-Mg alloys on mechanical properties**

<b>Alloy</b>	<b>Tensile Strength (MPa)</b>	<b>Hardness (HB)</b>	<b>Elongation (%)</b>
Alloy 1 (1% Cu)	152	45	12.7
Alloy 2 (3% Cu)	290	85	7.5
Alloy 3 (4.5% Cu)	360	105	5.4
Alloy 4 (6% Cu)	402	118	2.5

**Table 2.6: Effect of Si content of Al-Cu-Si-Mg alloys on mechanical properties**

<b>Alloy</b>	<b>Tensile Strength (MPa)</b>	<b>Hardness (HB)</b>	<b>Elongation (%)</b>
<b>Alloy 1 (0.5% Si)</b>	<b>152</b>	<b>45</b>	<b>12.7</b>
<b>Alloy 2 (7% Si)</b>	<b>164</b>	<b>65</b>	<b>7.2</b>
<b>Alloy 3 (12% Si)</b>	<b>180</b>	<b>90</b>	<b>3.0</b>
<b>Alloy 4 (18% Si)</b>	<b>200</b>	<b>102</b>	<b>1.1</b>

Raised tensile strength and hardness (due to precipitation hardness) are accompanied with a loss in elongation when Cu concentration is increased. Elongation increases and decreases with increased Si concentration, hardness, and tensile strength. A good combination of Cu and Si content is necessary for optimal mechanical characteristics.

### **2.5.7 Chemistry/Property relationships in AA206 alloy**

Sigworth and Major have explained that 206 T4 combinations will appear to exceed the 7 percent stretching requirement, now recognized as the standard for car suspension components, by providing enough ductility. The effects on mechanical properties of Fe and Si were also found to be not high in T4 mood. The achievement of high ductility with 206 combinations can present a more important challenge in the T7 mood. They proposed to achieve >7 percent of T7 mood pollution and rapid cementation rates

The following cycles of thermal treatment T4 and T7 are given [2]:

T4: Ramp from room temperature to 480°C in 1 hour  
Continue ramping from 480°C to 495°C in 1/4 hour  
Hold at 495°C for 2 hours  
Ramp from 495°C to 528°C in 1/2 hour  
Hold 10 hours at 528°C

Quench into 65°C water

Wait 7 days before mechanical testing

- T7: Solution treatment is same as outlined in above T4 schedule  
After the 65°C water quench hold 24 hrs at room temperature  
Age 4 hrs at 200°C  
Cool to room temperature in still air.

### 2.5.8 Mechanical properties of B206 alloy

Sigworth and Major details the mixable characteristics of B206 amalgam and the most commonly used combinations of aluminum (Fig. 2.9). The B206 amalgam has mechanical characteristics prevalent in common combinations of Al-Si-Mg and Al-Si-Cu.

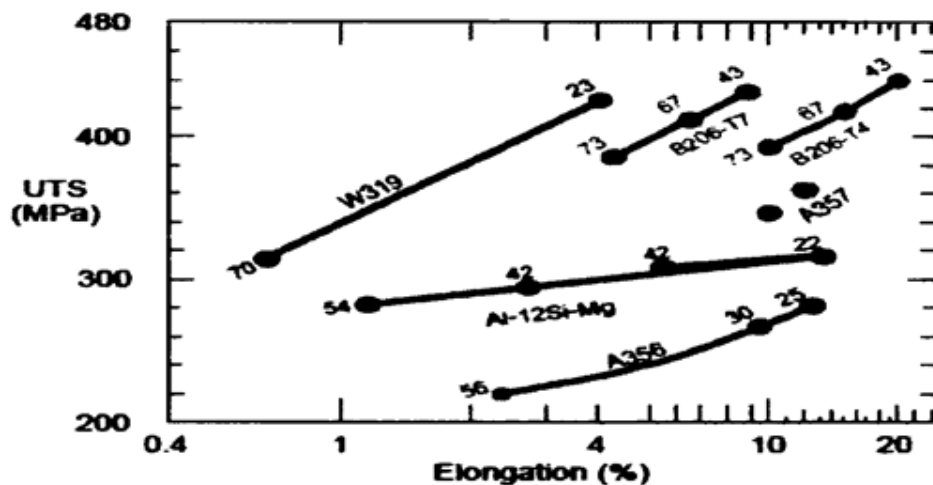


Figure 2.9: Range of mechanical properties in five aluminum casting alloys

### 2.6 Corrosion of Al-Cu alloys

Alloys in aqueous situations are prone to localized erosion, since they generally consist of significant amounts of constituent particles that are different from the grid and form local galvanic cells on metal surface promptly. In a variety of mechanically imperative metallic materials, the intergranular erosion occurs among the distinctive shapes of



localized erosion, such as combinations of aluminum with severe levels of different erosions.

Fells and grenary erosion can become a potential location for beginning of breaks, which occurs in catastrophic deception by declining erosion or erosion. In general, a near-by breakdown of the detached film on the surface results in both pitting and intergranular erosion, yet the exact instrument of the breakdown is not clear. The attacks on intermetallic particles or grain are referred to as settings, whereas an attack confined to grain borders is referred to as intergranular erosion. The setup and intergranular forms of erosion seem to be exceptionally comparative in nature from an electrochemical point of view.

Corrosion is often complicated in nature when multiphase amalgams are considered. The presence in the al-Cu amalgam of the 9-AI2Cu stage made it difficult to get its corrosion. The amalgam Al-Cu can be a multi-phase amalgam with an AI framework and acceleration of 0- AI2Cu. The a-AI system begins to dissolve and the six phase (since it contains Cu) acts as a respectable phase within the proximity of the destructive medium. Therefore, a greater volume distribution of the stage 6-AI2Cu within the combination will give the mixed potential hypothesis greater (less active) corrosion potentials

Agarwala and Murty [have carried out a controlled AI-4.5Cu alloy potential erosion test in 3.5% NaCl. The specific disintegration of the amatrix phase in the anodic potentially -0,300 to -0,500 V was detailed. In any case, since the associated polarization potential was negative, there was a shift in corrosion. A-matrix premises must be safe constantly and corrosion has begun in the 6-AI2Cu stage. At -0,900 V, the accelerating corrosion of the phase has happened, with the copper-rich amatrix phase being disintegrated.

### **2.6.1 Immersion testing of heat treated 201 alloys in NaCl/H2O2 corrosive solution**

201 heat-treated combination of copper ( 4.85%) and silver (0.72%) is vulnerable to erosion splitting as major alloying elements. As a work of the microstructuring conditions, Misra and Oswalt detailed their intergranular erosion defense and linked it to precipitation energy in the preparations for maturation. They performed intergranular tests on the treatment and disposition of aged specimens by drenching them for six hours in the arrangements for sodium chloride-hydrogen peroxide. In terms of the depth of corrosion infiltration within the local grain border, the severity of intergranular erosion was measured. Intergranular in 201 combinations attack increases

with increment in counterfeit maturing time (0 to 8 hrs), comes to a greatest at 8 hrs, and at that point diminishes when overaged (allude to Fig.2.10).

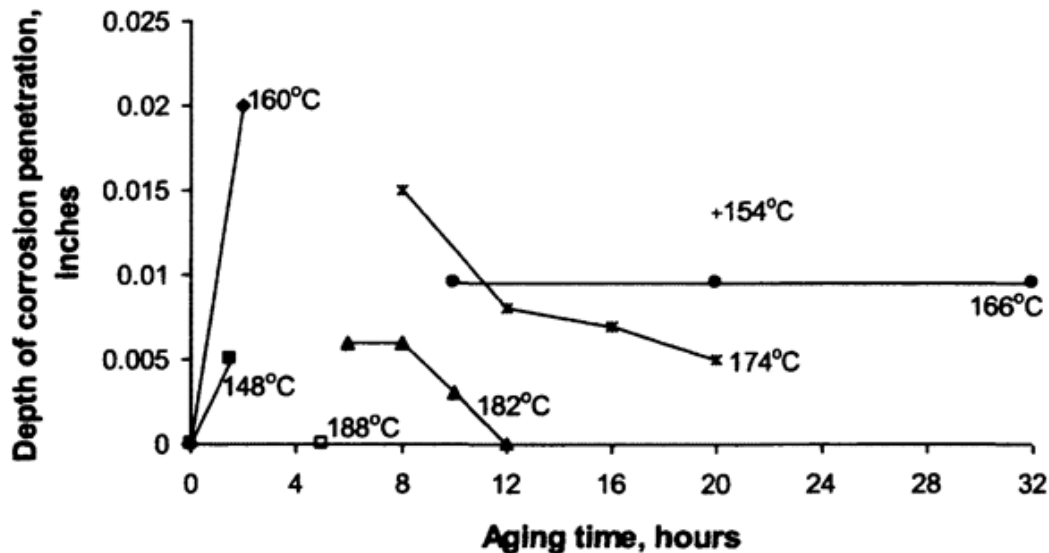


Figure 2.10: Intergranular corrosion characteristics of aged 201 alloy

## 2.6.2 Stress corrosion cracking

When exposed simultaneously to specific situations (water steam, watery systems, natural fluids and fluids) and stresses of appropriate size, aluminum and its alloys can fall apart by breaking along grain borders. Characteristically intergranular, stress-corrosion splitting in aluminum combinations. An intergranular (intercrystalline) corrosion may be a specific grain border attack or nearby districts without a significant grain attack. Intergrained erosion is caused by possible contrasts between the local grain boundary and the adjacent grain systems

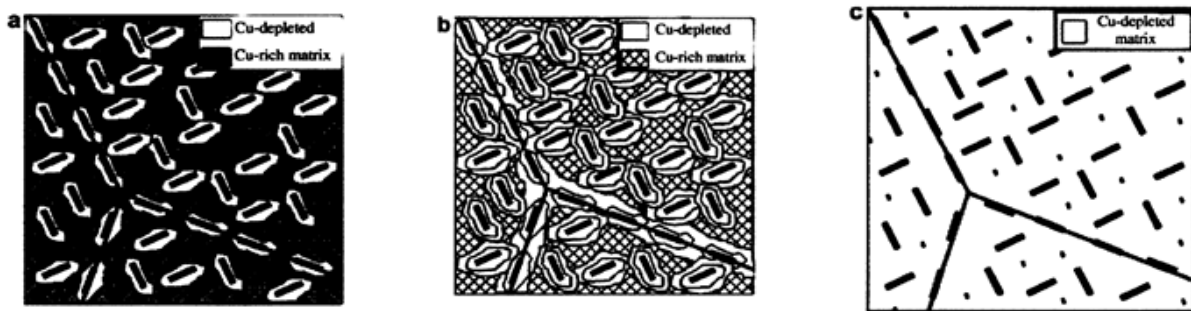
### 2.6.2.1 Intergranular stress corrosion behaviour in Al-Cu-Mg-Ag alloys

Al-Cu-Mg-Ag alloys can provide advanced mechanical properties, but often have localization erosion and intergranular erosion sensitivity. The proximity of the grain border increases the indefensibility of these alloys into intergranular localized attacks more often than not. In Al alloys, the strongly arranged copper substance directly concerns also setting Cu's potential. The expansion of Cu content within the strong



arrangement increases the framework's definition and reassessment potential. The phase of the  $\text{Al}_2\text{CuMg}$  increases in size and thickness but its composition remains unchanged when overage. The proximity of  $\text{Al}_2\text{CuMg}$  to grain borders cannot therefore easily clarify the impotence of inter-granular erosion splitting.

In over-ages T8+ conditions (e.g. T8 moods (crest aged) followed by an extra warm 5000 h introduction at a temperature of  $107.2^\circ$ ), small and in particular on progression to resistance to intergranular stress corrosion in two Al-Cu-Mg-Ag alloys in over-ages T8+ conditions (e.g. T8 moods (crest aged) followed by an extra warm 5000 h introduction at a temperature of  $107.2^\circ$ ) (Al-5.0 percent Cu-0.8 percent Mg-0.5 percent Ag and Al-5.4 percent Cu-0.5 percent Mg-0.5 percent Ag). To simulate the possible time for introducing service circumstances with hot temperatures, a 5000-hour heating operation at  $107.2^\circ\text{C}$  was carried out. According to the Al-Cu phase chart, Cu in an Al lattice has a balance solvency of 5.65 percent at  $548^\circ\text{C}$ , less than 1% at  $200^\circ\text{C}$ , and a little less than 0.1 percent at  $107.2^\circ\text{C}$ . This T8+ situation causes dispensing due to a difference in Cu concentration between the grain boundaries and the grain interiors. As a result of the leveling of Cu concentration, a unique disintegration pathway at grain borders known as intergranular push erosion breaking is suppressed.



**Figure 2.11: (a) It appears under-aged in which the border has the strongest differentiation of concentration among the Cu-depleted areas and the**

supersaturated (highest Cu in the framework), (b) It appears that the Cu-depletion zones around the Cu-containing accelerates begin to cover not only within the boundaries, but also within the grids. (c) Speaks of a scheme of the matrix Cu-depleted, which is linked to a remarkably old condition (lowest Cu in framework). The specific Cu drained route is eliminated along the grain borders. More nano-scale speeds frame during the long maturing times. There is a clear decline in the copper sum in strong arrangement due to an increase in the number of accelerated nanoscales over long maturation times.

#### 2.6.2.2 Pitting corrosion of Al-Cu alloys

Fixing of aluminum amalgams is the main common erosion attack. The pits form on the uncovered aluminum, new, or salt-water or other impartial electrolytes at localized discontinuities in the oxide film. The shape of a pit is generally hemispherical from shallow, subduous depression to hollow and round depressions. The defensive detached film is broken off by aggressive anions, such as the halides, and can lead to disastrous fabric disappointment. Intergranular erosion with setting erosion may occur. The amalgams of the 2xxx system are the least safe to pit. Diminishing the aggressiveness of the devastating system (for example, concentration of chloride ions, specialists in oxidation, temperature) can anticipate an attack.

Combinations of aluminum containing an enormous amount of copper are less resistant to corrosion than other. Kim and Buchheit [uncovered in the 0.1 M NaCl array for high imperfection Al, Al-0.2Cu and Al-2.0Cu. They decided that copper increases in aluminum (up to 2.0 Wt percent) decrease hindrance, as copper remains in a strong position. The potential to set up with the increase of copper in a combination can be seen as an increase.

### CONCLUSIONS

An extremely intergranular corrosion resulted in a heterogeneous Cu-containment dissemination which accelerates at the grain limits of the as-cast test. Moving 3-stage ST & 2-stage AA drugs will result in better precipitated aluminum copper than 2-stage ST & 1-stage AA drugs.

Normal maturation (NA) + Counterfeit maturation (AA) tests showed higher elongation than the false-maturation test without NA. Natural tests of mid-range erosion are, however, helpless. Therefore, NA harms both the misleadingly developed one and two-step B206 testing with respect to erosion and gravity.

Temperature is high The erosion resistance of the B206 combination was carried out by AA. Tall AA temperatures reduce the depth of intergranular erosion both in initial arrangements (2-step ST, NA (0 & 24 hours), 1-step AA) and in the third series (3-step ST, NA (0 & 24 hours), 2-step AA) for warm medicine. Third arrangement warm and longer medicines, to begin with step AA-time, dispose of intergranular erosion. NA ripening tests are susceptible to both erosion and intergranular erosion in the third arrangement.

For the 3-stage ST&2-stage ST & 1-stage AA, comparison of erosion shows that a 2-stage AA treatment seems to give way to better control of the carrot and gravity than a 1-step AA. Medicines leading to zones on the grain frontier and greater intermetallic accelerates along the grain frontiers, which have been subjected to intergranular attacks. High AA Temperature and longer AA holding time will decrease the amount of intermetal accelerations jointly decrease accelerating free zones near grain limits within the grain limits. This is why the O-D-8 test is highly erosion resistant.

Erosion and gravity modes derived from unmachined testing are comparable to those of the machined test tests because the destructive solution and forcible chemical pre-treatment of the inactive layer with the use of concentrated HN 03 for erosion testing were used earlier.

There were exceptional extensions of the current arrangement for warm medicines (2-stage ST, NA (72-hrs), 1-stage AA) which went from 4 to 20 per cent. The prolongation of warm medicines in the second and third arrangements is decreased by high AA and longer AA hold time. Whatever the case, high AA and longer AA temperatures offer superior resistance to erosion. High erosion resistance + high lengthenings for high AA temperatures and longer AA times are difficult to achieve thus.

## RECOMMENDATIONS

The type and shape of accelerates on the grain borders and grain network determine the resistance to intergranular/pitting erosion. To determine the nature and distribution of Cu containing intermetallic accelerators on grain boundaries and in the grain structure, more SEMTTEM study is required.

Future maturing medications should be designed to regulate the amount and distribution of Cu-containing intermetallic accelerates without compromising ductility or erosion resistance.

## REFERENCES

1. Sigworth, G.K. and Major, J.F. Factors influencing the mechanical properties of B206 alloy castings, in TMS Light Metals 2006. 2006. p. 795-799 .
2. Major, J.F. and Sigworth, G.K. Chemistry/Property relationships in AA206 alloys. AFS Transactions, 2006. Paper 06-209: p. 12.
3. Tresla, E., Aluminum Casting Alloys and Properties. AFS Transactions, 1964. 72: p. 840-849.
4. Duralumin: Retrieved on 7/23/2008. [cited; Available from: <http://en.wikipedia.org/wiki/Duralumin>.
5. Sigworth, G.K. and Dehart, F. Recent Developments in the High Strength Aluminum-Copper Casting Alloy A206. AFS Transactions, 2003.11: p. 341-354.
6. Davies, V.d.L., The influence of grain size on hot tearing. The British Foundryman, April 1970. 63: p. 93-101.
7. Pacz, A., Aluminum Alloy Casting and Process of Making the Same. 31 May 1932: U.S Patent, No. 1,860,947.
8. Sicha, W.E. and Boehm, R.C. Effect of titanium on grain size and tensile properties on an Aluminum-4.5% Copper (No. 195) Casting Alloy. AFS Transactions, 1948. 56: p. 398-409.
9. Fasoyinu, F.A., Thomson, J.P., Sahoo, M., Burke, P., Weiss, D. Permanent Mold Casting of Aluminum Alloys A206.0 and A535.0. AFS Transactions, 2007. 115, Paper 07-095(2).
10. Wannasin, J., Schwamm, D., Yurko, J.A., Rohloff, C., Woycik, G. Hot Tearing Susceptibility and Fluidity of Semi-Solid Gravity Cast Al-Cu Alloy. Diffusion and Defect Data Pt. B: Solid State Phenomena, 2006.116-117: p. 76-79.

11. Sigworth, G.K., High Strength casting alloys for automotive applications, in TMS Annual Meeting. 2002. p. 133-140.
12. Dimitrov, N., Mann, J.A., and Sieradzki, K. Copper redistribution during corrosion of aluminum alloys. Journal of The Electrochemical Society, 1999. 146 [1]: p. 98- 102
13. Van Horn K. R. Aluminum:vol I: Properties, physical metallurgy and phase diagrams. 1967, Metals park, Ohio, USA: ASM Handbook. 425.
14. Deshpande, J.U., The Effect of Mechanical Mold Vibration On the Characteristics of Aluminum Alloys, in Manufacturing Engineering. 2006, Worcester Polytechnic Institute, p. 113.
15. Davis, J.R. and Associates, Aluminum and Aluminum alloys. 4 ed. 1993: ASM International. 784.
16. Vadim S. Zolotarevsky, Nikolai A. Belov, and Glazoff, M,V. Casting Aluminum Alloys. 2007: Elsevier,. 544.
17. Chitty, A., The effect of Silicon, Manganese, and Iron on the ageing characteristics of Aluminium-